6 Carbon

Group 4 - The Crystallogens

THE GROUP 4 ELEMENTS ARE A MIX OF METALS, METALLOIDS & NON-METALS, AND SHOW CONSIDERABLE DIVERSITY IN THEIR PROPERTIES

14 Si Silicon

32 Ge Germanium

> 50 Sn Tin

82 Pb Lead

114 FlFlerovium

MELTING POINTS

GERMANIUM

937°C

TIN

232°C

LEAD

327°C

NUMBER OF ALLOTROPES OF CARBON

INCLUDING DIAMOND, GRAPHITE & BUCKMINSTER FULLERINE

SILICON IS THE SECOND MOST ABUNDANT **ELEMENT IN THE EARTH'S CRUST** ΑT

25.7%

GROUP 4 ELEMENTS FOUR VALENCE ELECTRONS

SILICON

1410°C

CARBON

3730°C

EXPOSURE LEAD CAN CAUSE LEAD POISONING

FLEROVIUM 0 0 ELEMEN DISCOVERED IN

¹¹²Sn ¹¹⁴Sn ¹¹⁵Sn ¹¹⁶Sn ¹¹⁷Sn



¹¹⁸Sn ¹¹⁹Sn ¹²⁰Sn ¹²²Sn ¹²⁴Sn

TIN IS THE ELEMENT WITH THE LARGEST NUMBER OF STABLE ISOTOPES

HYDRI GE

ЕΗ

TETRAHALIDES WITH HALOGENS

AND A VARIETY OF OXIDES WITH OXYGEN CARBON OF THE **HUMAN BODY**

IT ALSO FORMS ORGANIC COMPOUNDS THE STUDY OF WHICH IS KNOWN ORGANIC CHEMISTRY

THE CRYSTALLOGENS



CARBON

DIAMONDS

CARBON DATING



SILICON

GLASS

ELECTRONICS



GERMANIUM

MICROSCOPE & CAMERA LENSES



TIN

TIN CANS



RADIATION SHIELDING

LEAD

LEADED PETROL